







Investigations on Fatigue of Li-ion batteries

HELMUT EHRENBERG

INSTITUTE FOR APPLIED MATERIALS – ENERGY STORAGE SYSTEMS (IAM-ESS)



KIT – University of the State of Baden-Wuerttemberg and National Research Center of the Helmholtz Association

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Introduction into Fatigue of Li-ion batteries: General aspects

Materials challenges and the cell level

in operando techniques for life-time studies:

Neutron tomography and diffraction on 18650-type batteries

Conclusion

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Fatigue: Degradation of performance with operation



Number of charge/discharge cycles or time

- Loss of capacity
- Increase of internal resistance

Voltage fade

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Failure modes: "Thermal runaway" Serious safety concerns at the EOL (end of life)





 $\text{Li}_{1-x}\text{CoO}_2$ is intrinsically unstable in the overcharged state (x > 0.5) $\text{Li}_{0.5}\text{CoO}_2 \xrightarrow{T \ge 200^{\circ}\text{C}} 1/2 \text{ LiCoO}_2 + 1/6 \text{ Co}_3\text{O}_4 + 1/6 \text{ O}_2$

Enhanced materials for safer operation (NCM, NCA)
Design freezing (~7 y in cars, ~10 y in planes)

Safety aspects: energy within a Li-ion battery





A charged Li-ion battery stores about the same energy as a stick of dynamite of the same size!



Fatigue: Multi length-scale complexity













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- All components suffer from "Ageing" & "Fatigue"
- Materials interactions: "Solid Electrolyte Interface/Interphase", SEI "Metal dissolution", "Loss of adhesion"



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Relative shift of the states-of-charge of cathode and anode results in capacity losses on full-cell level (example: NCA against graphite)

Electrode-specific contributions to fatigue





- Overvoltages from the negative electrode result in a higher level of Liextraction from the positive electrode and, therefore,
- more pronounced fatigue.



Cyclovoltamograms of full cells with reference electrode (NCA against graphite) in combination with half-cell data (NCA and graphit against Li).

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Electrode-specific contributions to fatigue





Aged cell	Capacity loss
Cathode	
Cell balancing	2,1 % ± 0,01 % ¹⁾
Fatigued	Capacity loss
Cathode	9,0 % ± 1,3 % ¹⁾
Cell balancing	10,0 % ± 2,0 % ¹⁾
Highly fatigued	Capacity loss
Cathode	28,5 % ± 1,3 % ¹⁾
Cell balancing	9,7 % ± 0,7 % ¹⁾

1) Maximum deviation within three experiments

in operando observation of anode excess







- More intermediate LiC₁₂ in the "fully" charged state due to fatigue, i.e.
- less Li in the anode in fatigued state due to "anode excess".



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Cycle life testing and fatigue analysis



- Commercial Li-ion cell of 18650-type (2600 mAh, 3.0 4.2 V)
 - "Fresh" single cycle
 - "Fatigued" at 25°C and 50°C (200, 400, 600, 800, 1000 cycles)
 - Cycling 3.0 4.2 V, CCCV, 1C



O. Dolotko, J. Electrochem. Soc. 159 (2012) A2082

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Cycle-life testing and fatigue analysis



Commercial Li-ion cell of 18650-type (2600 mAh, 3.0 - 4.2 V)











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in operando techniques for life-time studies



- Pronounced materials interactions: real battery conditions essential
- Specific degradation mechanisms:
 - Correlation between effects and cause
 - Influence of load profile
- Proceeding of fatigue
- Cell design concepts



High-energy synchrotron & neutron radiation

- Non-destructive methods
- Penetration capability
- Time- and spatial resolution
- Detailed information (minor changes)
- Sensitivity to light elements (H, Li, C, O)
- Complementary *post mortem* analysis





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Neutron radiography and tomography



Visualisation by assignment of absorption levels to a false-color scheme



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Selected examples



X-ray tomography reveals shifts with SOC



Selected examples



Neutron tomography: electrolyte level changes during charge/discharge



Selected examples



Combined neutron tomography & diffraction



Fresh Cell

Fatigued Cell (1000 cycles)

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see "Highlights" @ http://www.iam.kit.edu/ess

Neutron Powder Diffraction





O. Dolotko, J. Electrochem. Soc. 159 (2012) A2082

Anode: Ratio of LiC₆ and LiC₁₂ in charged state





Phase fraction of LiC_6 is lower for the fatigued 25 °C cell, while the LiC_{12} value is higher.

Reduction of lithium inside the anode: $25 \,^{\circ}C \rightarrow 26 \,^{\circ}\%$ $50 \,^{\circ}C \rightarrow 15 \,^{\circ}\%$

More intermediate LiC₁₂ in the "fully" charged state due to fatigue, i.e.

- less Li in the anode in fatigued state due to "anode excess" by shift of the voltage window.
- Elevated temperature: less overvoltage and less fatigue.

O. Dolotko, J. Electrochem. Soc. 159 (2012) A2082





- Li occupation in the discharged state higher at 50°C.
- Safety problems due to low Li-content in the carged state also at higher temperature!

O. Dolotko, J. Electrochem. Soc. 159 (2012) A2082

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Lattice strain in layered oxides LCO, NCM, NCA Phase behaviour of LiNi_{0.8}Co_{0.2}O₂ LiNi_{1/3}Co_{1/3}Mn_{1/3}O₂ Phase 1 2.86 2.86 Phase 2 2.85 °**∀** 2.85 ∕ 8 2.84 Å / 2.84 a 2.83 2.82 2.83 2.81 14.5 14.4 14.4 •**ح**^{14.}[≀] ∪ 14.0 14.2 •**∢** 14.3 ບ 14.2 14.1 Oxidation of oxygen 13.8 14.0 100 101 Vol / v 99 98 Vol / $Å^3$ 98 96 97 94 charge 0.5 0.15 0.5 0.85 0.5 0.0 0.5 1.0 1.0 *x* Li in Li_xNi_{0.8}Co_{0.2}O₂ *x* Li in $Li_x Ni_{1/3} Co_{1/3} Mn_{1/3} O_2$ discharge

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Lattice strain and microstructure: NCM





SEM images of as-prepared $Li(Ni_{1/3}Co_{1/3}Mn_{1/3})O_2$ powder

N. Kiziltas-Yavuz et al., *Electrochim. Acta.* **113** (2013) 313

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Lattice strain and microstructure: NCM



SEM images of Li(Ni $_{1/3}$ Co $_{1/3}$ Mn $_{1/3}$)O $_2$ after cycling between 3.0 and 4.2 V



N. Kiziltas-Yavuz et al., *Electrochim. Acta.* **113** (2013) 313

Cathode fatigue



- Positive electrode materials are the only source of exchanged Li.
- Most serious limitation for the energy density of a cell.
- Restricted stability with respect to the Li-content (except LiFePO₄).
- Details of the electronic structure play a crucial role for
 - cell voltage,
 - electronic conductivity,
 - stability range, band overlap: oxidation of oxygen,
 - activitiy of specific ions (energy levels)
 - and more...
- Structure and composition gradients within one particle/crystallite, for example a NiO-type surface layer in NCA.
- One-phase or two-phase mechanisms.
- Significant structural distortions and volume changes.
- High probability for cracks, especially in blend cathodes, working over a broad voltage range.









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Conclusions



- Fatigue of Li-ion batteries has to be addressed on cell level.
- Capacity losses and internal resistance are determined by materials and balancing.
- Materials interactions are very pronounced in batteries.
- No individual optimization of materials, only within specific cells!
- Dedicated methods are needed to investigate fatigue under real operation conditions.
- Data analysis is a specific challenge.
- Complementary methods are essential.
- Probably not all basic physical parameters can be measured.
- Complex functionalities have to be considered
- Too many degrees of freedom: only knowledge-based approaches are promising.

Acknowledgement





Natalia Bramnik Susana Darma Oleksandr Dolotko Markus Herklotz Karin Kleiner Michael Lang Martin Mühlbauer Kristian Nikolowski Lars Riekehr Anatoliy Senyshyn Florian Sigel

&

all SFB members

